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Xiao<sup>b</sup> and Long-Guan Zhu<sup>a\*</sup><sup>a</sup>Department of Chemistry, Zhejiang University, Hangzhou 310027, People's Republic of China, and <sup>b</sup>School of Chemistry and Materials Science, Wenzhou University, Wenzhou 325027, People's Republic of China

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## Key indicators

Single-crystal X-ray study

T = 295 K

Mean  $\sigma(\text{C}-\text{C}) = 0.006 \text{ \AA}$ 

R factor = 0.065

wR factor = 0.145

Data-to-parameter ratio = 12.6

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.Aquabis(1,10-phenanthroline)(3-sulfonato-  
benzoato)manganese(II) tetrahydrate

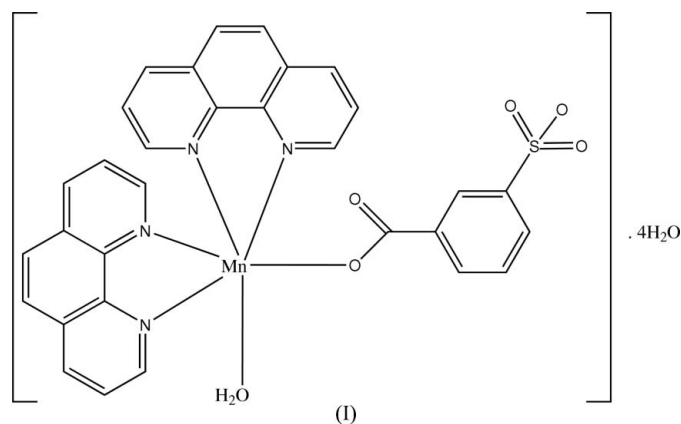
The title compound,  $[\text{Mn}(\text{C}_7\text{H}_4\text{O}_5\text{S})(\text{C}_{12}\text{H}_8\text{N}_2)_2(\text{H}_2\text{O})] \cdot 4\text{H}_2\text{O}$ , which was obtained by the reaction of  $\text{MnCl}_2$  with sodium hydrogen 3-sulfobenzoate and 1,10-phenanthroline, is a monomer. The coordination polyhedron of the Mn atom is a distorted octahedron.  $\pi-\pi$  Interactions between neighboring monomers and extensive hydrogen bonding consolidate the stability of the crystal structure.

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## Comment

Recently, 12 3-sulfobenzoate compounds have been synthesized in our laboratory and by other groups. These compounds include one organic complex (Adrabsinska *et al.*, 2001) and 11 metal complexes in which the metal atoms include lead (Ying & Mao, 2004; Ma *et al.*, 2005; Zhu & Xiao, 2005), barium (Gao *et al.*, 2005), zinc (Li *et al.*, 2005; Zhang & Zhu, 2005), cadmium (Chen *et al.*, 2005; Miao & Zhu, 2006) and copper (Cai *et al.*, 2005; Cai & Zhu, 2005; Miao *et al.*, 2005). As part of an investigation on 3-sulfobenzoate metal complexes, a manganese 1,10-phenanthroline compound has been prepared, (I).



Compound (I) is a monomer, in which the Mn atom adopts a distorted octahedral geometry completed by four N-atom donors from two 1,10-phenanthroline, one O atom from one 3-sulfonatobenzoate, and one O atom from a water molecule (Fig. 1 and Table 1). The 3-sulfonatobenzoate ligand coordinates to the Mn atom in a monodentate manner *via* one O atoms of the carboxylate group. The carboxylate group is nearly coplanar with the benzene ring to which it is attached, forming a dihedral angle of  $2.1(3)^\circ$ .

In neighboring monomers, there is a strong  $\pi-\pi$  interaction between 1,10-phenanthroline ligands, with a centroid-to-centroid distance of  $3.497(2) \text{ \AA}$ . Extensive hydrogen bonding between water molecules, and between water molecules and

sulfonyl groups, give rise to a three-dimensional network (Fig. 2).

**Experimental**

A mixture of MnCl<sub>2</sub> (0.032 g, 0.25 mmol), sodium hydrogen 3-sulfo-benzoate (0.056 g, 0.25 mmol) and 1,10-phenanthroline (0.051 g, 0.25 mmol) in aqueous solution (20 ml) was stirred for 1 h and then filtered. The resulting solution was allowed to evaporate at room temperature. After 2 d, pale-yellow crystals were filtered off and washed with water.

*Crystal data*

[Mn(C<sub>7</sub>H<sub>4</sub>O<sub>5</sub>S)(C<sub>12</sub>H<sub>8</sub>N<sub>2</sub>)<sub>2</sub>·(H<sub>2</sub>O)]·4H<sub>2</sub>O  
*M<sub>r</sub>* = 705.59  
 Monoclinic, *C*2/*c*  
*a* = 33.549 (7) Å  
*b* = 8.1406 (15) Å  
*c* = 23.584 (4) Å  
 β = 90.032 (8)°

*V* = 6441 (2) Å<sup>3</sup>  
*Z* = 8  
*D<sub>x</sub>* = 1.455 Mg m<sup>-3</sup>  
 Mo Kα radiation  
 μ = 0.54 mm<sup>-1</sup>  
*T* = 295 (2) K  
 Block, pale yellow  
 0.33 × 0.21 × 0.14 mm

*Data collection*

Bruker APEX area-detector diffractometer  
 φ and ω scans  
 Absorption correction: multi-scan (SADABS; Bruker, 2002)  
*T<sub>min</sub>* = 0.843, *T<sub>max</sub>* = 0.929

16479 measured reflections  
 5733 independent reflections  
 5039 reflections with *I* > 2σ(*I*)  
*R<sub>int</sub>* = 0.033  
 θ<sub>max</sub> = 25.1°

*Refinement*

Refinement on *F*<sup>2</sup>  
*R* [*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.065  
*wR* (*F*<sup>2</sup>) = 0.145  
*S* = 1.16  
 5733 reflections  
 454 parameters  
 H-atom parameters constrained

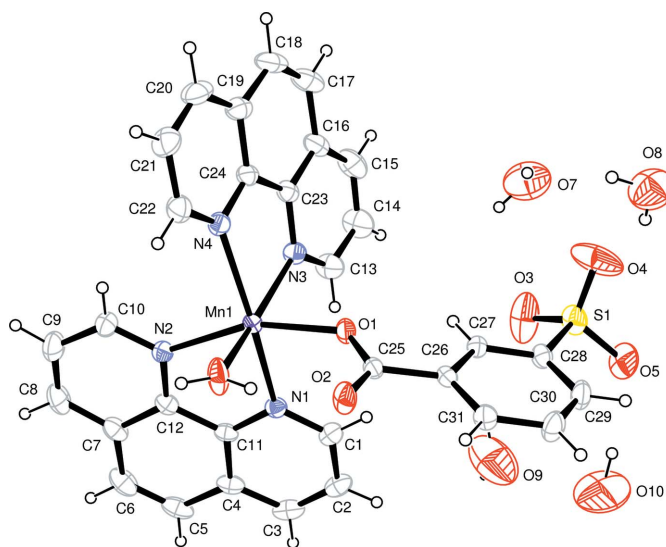
*w* = 1/[σ<sup>2</sup>(*F<sub>o</sub>*<sup>2</sup>) + (0.0531*P*)<sup>2</sup> + 13.5505*P*]  
 where *P* = (*F<sub>o</sub>*<sup>2</sup> + 2*F<sub>c</sub>*<sup>2</sup>)/3  
 (Δ/σ)<sub>max</sub> < 0.001  
 Δρ<sub>max</sub> = 0.43 e Å<sup>-3</sup>  
 Δρ<sub>min</sub> = -0.48 e Å<sup>-3</sup>

**Table 1**

Selected geometric parameters (Å, °).

Mn1—O1	2.095 (2)	Mn1—N1	2.295 (3)
Mn1—O6	2.165 (3)	S1—O4	1.411 (4)
Mn1—N2	2.245 (3)	S1—O3	1.431 (4)
Mn1—N3	2.258 (3)	S1—O5	1.441 (3)
Mn1—N4	2.270 (3)		
O1—Mn1—O6	85.45 (10)	N3—Mn1—N4	73.34 (11)
O1—Mn1—N2	160.12 (10)	O1—Mn1—N1	89.09 (10)
O6—Mn1—N2	88.33 (11)	O6—Mn1—N1	99.37 (12)
O1—Mn1—N3	86.42 (10)	N2—Mn1—N1	73.31 (10)
O6—Mn1—N3	164.10 (12)	N3—Mn1—N1	94.13 (10)
N2—Mn1—N3	103.66 (11)	N4—Mn1—N1	162.41 (11)
O1—Mn1—N4	102.08 (10)	O4—S1—O3	113.3 (3)
O6—Mn1—N4	95.06 (12)	O4—S1—O5	112.8 (3)
N2—Mn1—N4	97.25 (10)	O3—S1—O5	111.4 (2)

H atoms bonded to C atoms were positioned geometrically and refined as riding, with C—H = 0.93 Å and *U*<sub>iso</sub>(H) = 1.2*U*<sub>eq</sub>(parent). The water H atoms were located in difference Fourier maps and



**Figure 1**

The asymmetric unit of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 40% probability level.

refined with restraints for O—H distances [0.85 (1) Å] and with *U*<sub>iso</sub>(H) = 0.08 Å<sup>2</sup>.

Data collection: *SMART* (Bruker, 2002); cell refinement: *SAINTE* (Bruker, 2002); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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